

## JEE Advanced - 1 | Paper - 2 | JEE 2024

Date : 29/10/2023

Maximum Marks: 180

Timing: 2:00 PM to 5:00 PM

Duration: 3.0 Hours

### General Instructions

- The question paper consists of 3 Subject (Subject I: **Physics**, Subject II: **Chemistry**, Subject III: **Mathematics**). Each Part has **four** sections (Section 1, Section 2, Section 3 and Section 4).
- Section 1** contains **4 Multiple Choice Questions**. Each question has 4 choices (A), (B), (C) and (D), out of which **ONLY ONE CHOICE** is correct.  
  
**Section 2** contains **3 Multiple Correct Answers Type Questions**. Each question has 4 choices (A), (B), (C) and (D), out of which **ONE OR MORE THAN ONE CHOICE** is correct.  
  
**Section 3** contains **2 Paragraphs**. Based on each paragraph, there are **TWO (02)** questions. The answer to each question is a **NUMERICAL VALUE**. If the numerical value has more than two decimal places, **truncate/round-off** the value to **TWO** decimal places  
  
**Section 4** contains **6 Non-Negative Integer Type Questions**. The answer to each question is a **NON-NEGATIVE INTEGER**.
- For answering a question, an ANSWER SHEET (OMR SHEET) is provided separately. Please fill your **Test Code**, **Roll No.** and **Group** properly in the space given in the ANSWER SHEET.

### Syllabus:

**Physics:** Electrostatics, DC Circuits, Capacitors, Magnetic Effect of Current, EMI, AC Circuits, EM Waves

**Chemistry:** Solid State, Theory of Solution, Chemical Kinetics, Electrochemistry, Surface Chemistry, Organic Halides, Organic Concepts, OCOC-I, II & III

**Mathematics :** Functions, Inverse Trigonometry, DC - I & II, IC - I & II, Statistics

Name of the Candidate (In CAPITALS) : .....

Roll Number : .....

OMR Bar Code Number : .....

Candidate's Signature : ..... Invigilator's Signature .....

**MARKING SCHEME**

**SECTION – 1 | (Maximum Marks: 12)**

- This section contains **Four (04)** Multiple Choice Questions.
- Each question has **FOUR** options (A), (B), (C) and (D). **ONLY ONE** of these four options is the correct answer.
- For each question, choose the option corresponding to the correct answer.
- Answer to each question will be evaluated according to the following marking scheme.  
*Full Marks* : +3 If **ONLY** the correct option is chosen.  
*Zero Marks* : 0 If none of the options is chosen (i.e. the question is unanswered).  
*Negative Marks* : -1 In all other cases.

**SECTION – 2 | (Maximum Marks: 12)**

- This section consists of **Three (03)** Questions. Each question has **FOUR** options. **ONE OR MORE THAN ONE** of these four option(s) is(are) correct answer(s).
- Answer to each question will be evaluated according to the following marking scheme:  
**Full Marks:** +4 If only (all) the correct option(s) is(are) chosen  
**Partial Marks:** +3 If all the four options are correct but **ONLY** three options are chosen  
**Partial Marks:** +2 If three or more options are correct but **ONLY** two options are chosen and both of which are correct  
**Partial Marks:** +1 If two or more options are correct but **ONLY** one option is chosen, and it is a correct option  
**Zero Mark:** 0 if none of the options is chosen (i.e. the question is unanswered)  
**Negative Marks:** -2 In all other cases.

**SECTION – 3 | (Maximum Marks: 12)**

- This section contains **Two (02)** Paragraphs. Based on each paragraph, there are **TWO (02)** questions. The answer to each question is a **NUMERICAL VALUE**.
- If the numerical value has more than two decimal places, **truncate/round-off** the value to **TWO** decimal places.
- Answer to each question will be evaluated according to the following marking scheme.  
*Full Marks* : +3 If **ONLY** the correct numerical value is entered in the designated place.  
*Zero Marks* : 0 In all other cases

**SECTION – 4 | (Maximum Marks: 24)**

- This section contains **SIX (06)** Questions.
- The answer to each question is a **NON-NEGATIVE INTEGER**
- Answer to each question will be evaluated according to the following marking scheme.  
*Full Marks* : +4 If **ONLY** the correct integer is entered;  
*Zero Marks* : 0 In all other cases.

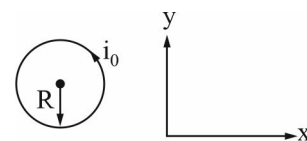
## SUBJECT I : PHYSICS

60 MARKS

## SECTION-1

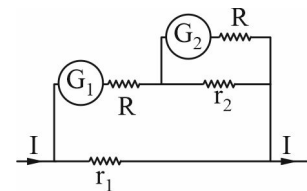
This section consists of 4 Multiple Choice Questions. Each question has 4 choices (A), (B), (C) and (D), out of which ONLY ONE CHOICE is correct.

1. A small current carrying ring having current  $i_0$  and radius  $R$  is kept in  $x$ - $y$  plane (the plane of paper) as shown in figure. Another current carrying small ring having radius  $r$  ( $r \ll R$ ) is kept at a large distance  $d$  from the centre of first ring in  $yz$  plane such that the centres of both rings lie on the  $z$ -axis and current in second ring is in anticlockwise direction as seen from +ve  $x$ . Find the torque acting on second ring due to the magnetic field of first ring.



- (A) Zero
- (B)  $\frac{\mu_0}{4} \frac{\pi i_0 R^2 \cdot r^2}{d^3} (-\hat{j})$
- (C)  $\frac{\mu_0}{2} \frac{\pi i_0 R^2 \cdot r^2}{d^3} (-\hat{j})$
- (D)  $\frac{\mu_0}{2} \frac{i_0 R^2 \cdot r^2}{d^3} (\hat{j})$

2. The diagram shows two galvanometers  $G_1$  and  $G_2$ . When current  $I = 1A$ , both  $G_1$  and  $G_2$  show full scale deflection. It is given that  $G_1$  shows full scale deflection for 10 mA and  $G_2$  shows full scale deflection for 1 mA. The values of  $r_1$  and  $r_2$  are : ( $G_1$  and  $G_2$  are of negligible resistances)



- (A)  $r_1 = \frac{R}{9}$  and  $r_2 = \frac{R}{90}$
- (B)  $r_1 = \frac{R}{90}$  and  $r_2 = \frac{R}{9}$
- (C)  $r_1 = \frac{2R}{9}$  and  $r_2 = \frac{2R}{90}$
- (D)  $r_1 = \frac{4R}{9}$  and  $r_2 = \frac{4R}{90}$

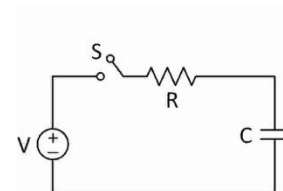
SPACE FOR ROUGH WORK

3. The intensity of an electric field depends on the co-ordinates  $x, y$  and  $z$  as follows  $\vec{E} = \frac{a(x\hat{i} + y\hat{j} + z\hat{k})}{(x^2 + y^2 + z^2)^{3/2}}$ .

The electrostatic energy stored between two imaginary concentric spherical shells of radii  $R$  and  $2R$  with centre at origin is:

- (A)  $\frac{4\pi\epsilon_0 a^2}{R}$  (B)  $\frac{2\pi\epsilon_0 a^2}{R}$  (C)  $\frac{\pi\epsilon_0 a^2}{R}$  (D)  $\frac{\pi\epsilon_0 a^2}{2R}$

4. Consider a simple RC circuit as shown in Figure. The voltage is first set to  $\frac{V_0}{3}$  and maintained for a charging time  $T \gg RC$ . Then the voltage is raised to  $\frac{2V_0}{3}$  without discharging the capacitor and again maintained for a time  $T \gg RC$ . The process is repeated one more time by raising the voltage to  $V_0$ .



Total energy dissipated across the resistance  $E_D$  in the process is :

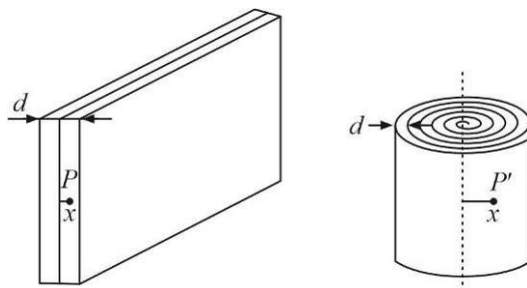
- (A)  $E_D = 3\left(\frac{1}{2}CV_0^2\right)$  (B)  $E_D = \frac{1}{2}CV_0^2$   
 (C)  $E_D = 3CV_0^2$  (D)  $E_D = \frac{1}{3}\left(\frac{1}{2}CV_0^2\right)$

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## SECTION-2

This section consists of 3 Multiple Correct Answers Type Questions. Each question has 4 choices (A), (B), (C) and (D), out of which **ONE OR MORE THAN ONE CHOICE** is correct.

5. For plane electromagnetic waves propagating in the +z-direction, which of the following combination(s) give the correct possible direction for  $\vec{E}$  and  $\vec{B}$  field, respectively?
- (A)  $(\hat{i} + 2\hat{j})$  and  $(2\hat{i} - \hat{j})$  (B)  $(-2\hat{i} - 3\hat{j})$  and  $(3\hat{i} - 2\hat{j})$   
 (C)  $(2\hat{i} + 3\hat{j})$  and  $(\hat{i} + 2\hat{j})$  (D)  $(\hat{j})$  and  $(-\hat{i})$
6. A non-conducting sheet of thickness  $d$  and large surface area contains a uniformly distributed charge of density  $\rho$  thorough out its volume. The electric field at a point  $P$  inside the sheet at a distance ' $x$ ' from the central plane is  $E_1$ . Now the sheet is rolled to form a large solid cylinder. Field at point  $P$  inside the cylinder at a distance ' $x$ ' from the axis is  $E_2$ . Then:



- (A)  $E_1 = \frac{\rho x}{\epsilon_0}$  (B)  $E_2 = \frac{\rho x}{2\epsilon_0}$  (C)  $E_1 = \frac{\rho}{2\epsilon_0}$  (D)  $E_1/E_2 = 1$
7. The current density in a wire of radius ' $a$ ' varies with radial distance ' $r$ ' as  $J = kr^2$ , where  $k$  is a constant. Choose the correct statements.
- (A) Total current passing through the cross section of the wire is  $I = \frac{\pi ka^4}{2}$   
 (B) Total current passing through the cross section of the wire is  $I = \frac{3\pi ka^3}{2}$   
 (C) The magnetic field at a radial distance  $r > a$  is  $B = \frac{\mu_0 ka^4}{4r}$   
 (D) The magnetic field at a radial distance  $r < a$  is  $B = \frac{\mu_0 kr^3}{4}$

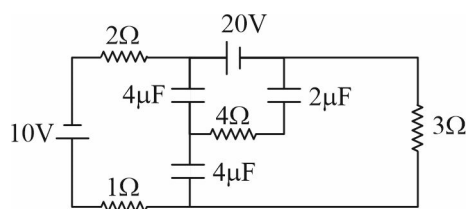
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### SECTION-3

This section consists of 2 Paragraphs. Based on each paragraph, there are TWO (02) questions. The answer to each question is a NUMERICAL VALUE. If the numerical value has more than two decimal places, truncate/round-off the value to TWO decimal places

#### Paragraph for Questions 1 - 2

The circuit shown in figure is in steady state. All the cells in the given circuit are ideal. Answer the following two questions:

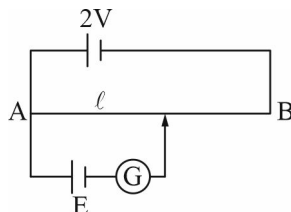


- Charge on  $2\mu F$  capacitor is \_\_\_\_\_  $\mu C$ .
- Ratio of energy stored in the two  $4\mu F$  capacitors is \_\_\_\_\_. (Give the ratio which is greater than one)

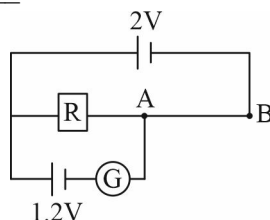
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**Paragraph for Questions 3 - 4**

Potentiometer is an instrument that can be used to measure the emf of a source without drawing any current from the source. As shown in the figure, the potentiometer wire AB of length  $10\text{m}$  and resistance  $20\Omega$  is connected to a battery of emf  $2.0\text{V}$ . Then, the emf of projected cell  $E = K\ell$ , where  $K$  is voltage drop per meter of potentiometer wire AB and  $\ell$  is the balancing length, when  $G$  the galvanometer shows null deflection [Battery is of negligible internal resistance].



3. A cell of emf  $1.2\text{V}$  gives null deflection at certain length  $x\text{m}$  of potentiometer wire and for another cell of emf  $E$ , the balancing length is found to be  $(x + 0.6)\text{m}$ . The emf of second cell then is \_\_\_\_\_ V.
4. Now a resistance box is introduced in the circuit as shown. The resistance in the box  $R$  is adjusted to get null deflection in the galvanometer so that the potential difference per metre of the potentiometer wire AB is  $1\text{mV}$ . The resistance  $R$  is \_\_\_\_\_  $k\Omega$ .



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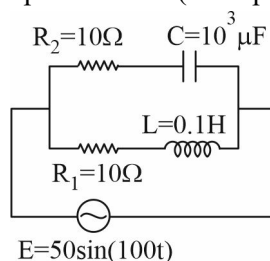
### SECTION-4

This section consists of 6 NON-NEGATIVE INTEGER Type Questions. The answer to each question is a NON-NEGATIVE INTEGER.

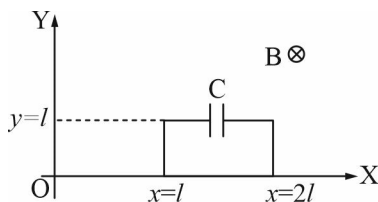
5. A charged particle is projected in magnetic field  $\vec{B} = 10\hat{k}$  from origin in  $x - y$  plane. The particle moves in a circle and just touches the line  $y = 5 \text{ m}$  at  $x = 5\sqrt{3} \text{ m}$ . Find the speed of particle (in m/s) given that its mass  $= 5 \times 10^{-5} \text{ kg}$  and charge  $= 1\mu\text{C}$ .

6. A plane electromagnetic wave,  $E = 100 \cos(6 \times 10^8 t + 4x) \text{ V/m}$ , is propagating in a medium. The dielectric constant of the medium is \_\_\_\_\_.

7. For the circuit shown in the figure, find the peak current (in ampere) through the source.



8. A square frame of wire connected to a capacitor is kept in a magnetic field which varies with position as well as time and is given as  $\vec{B} = \alpha x t (-\hat{k})$  as shown in figure, where  $\alpha$  is a constant. The charge on the capacitor at any time is  $\frac{n}{2} C \alpha l^3$ . Find  $n$ .



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9. A region in space contains a total positive charge  $Q$  that is distributed spherically such that the charge density is given by

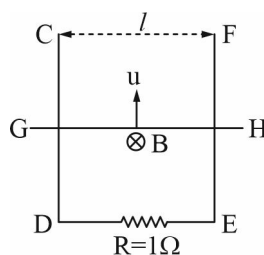
$$\begin{aligned}\rho(r) &= \rho_0 \text{ for } r \leq \frac{R}{2} \\ &= 2\rho_0 \left(1 - \frac{r}{R}\right) \text{ for } \frac{R}{2} \leq r \leq R \\ &= 0 \text{ for } r \geq R\end{aligned}$$

It is found that an electron executes simple harmonic motion about the center  $r = 0$  of the distribution with time period  $T = KR^{x/2}$ , if and only if the amplitude is less than  $R/2$ .

Find the value of  $x$ .  $\left[ K^2 = \left( \frac{15\pi\epsilon_0 m}{8eQ} \right) \right]$

10. In the figure, CDEF is a fixed conducting smooth frame in a vertical plane. A conducting uniform rod GH of mass  $m = 3\text{kg}$  can move vertically and smoothly without losing contact with the frame. GH always remains horizontal. It is given velocity  $u = 30\text{ms}^{-1}$  upwards and released. Taking the acceleration due to gravity as  $g$  and assuming that no resistance is present other than  $R = 1\Omega$ , find out time (in S) taken by rod to reach the highest point. Round-off your answer to nearest integer.

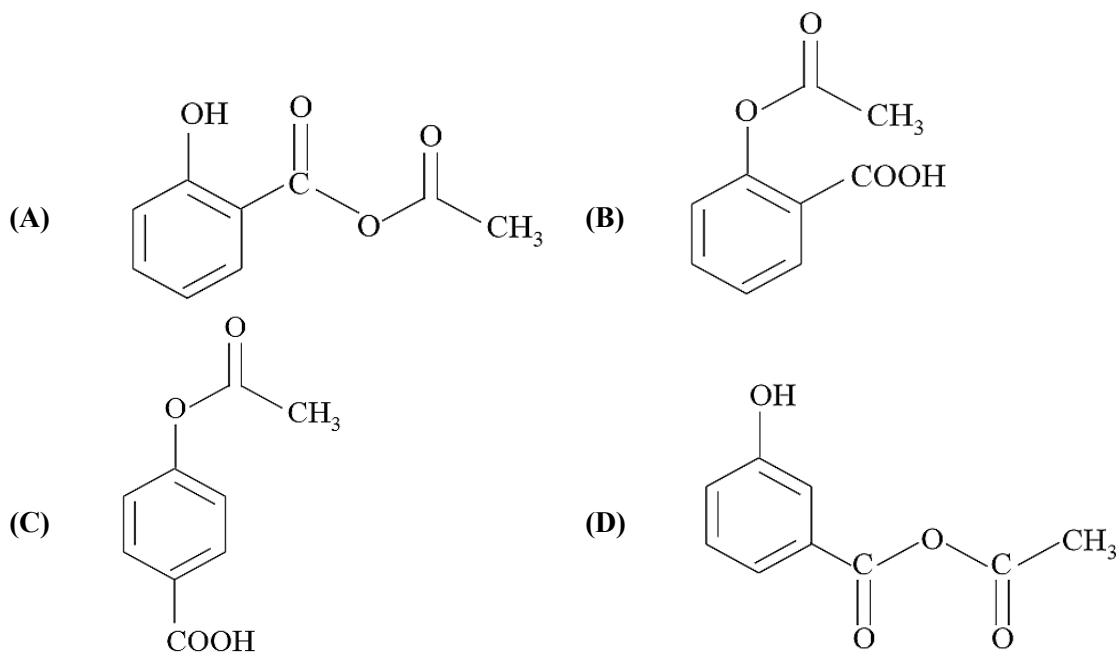
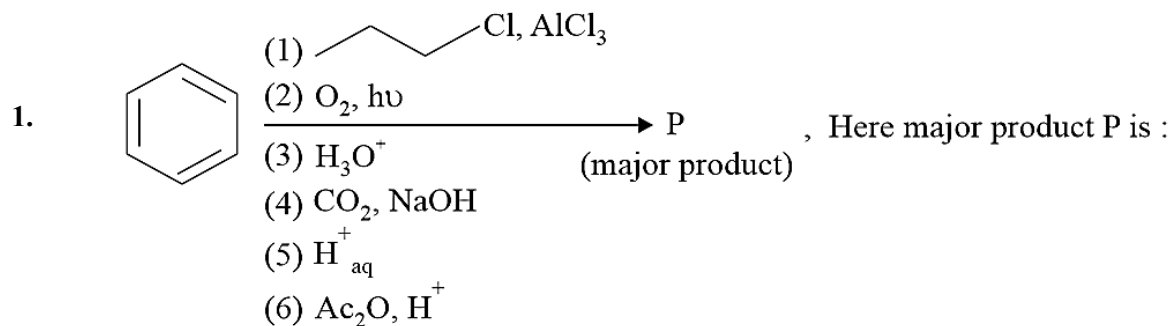
(Given  $B = 1\text{T}$ ,  $g = 10\text{m/s}^2$ ,  $l = 1\text{m}$ ,  $\ln(2) = 0.7$ )



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**SUBJECT II : CHEMISTRY****60 MARKS****SECTION-1**

This section consists of 4 Multiple Choice Questions. Each question has 4 choices (A), (B), (C) and (D), out of which **ONLY ONE CHOICE** is correct.

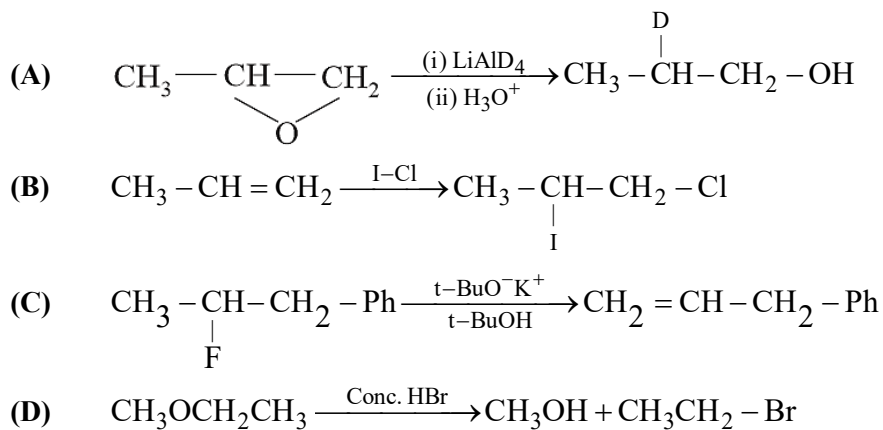


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2. A closed vessel with rigid walls contains 1 mol of  ${}^{238}_{92}\text{U}$  and 2 mol of air at 298 K. Considering complete decay of  ${}^{238}_{92}\text{U}$  to  ${}^{206}_{82}\text{Pb}$ , the ratio of the final pressure to the initial pressure of the system at 298 K is \_\_\_\_\_.

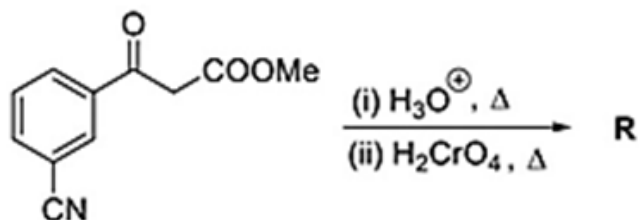
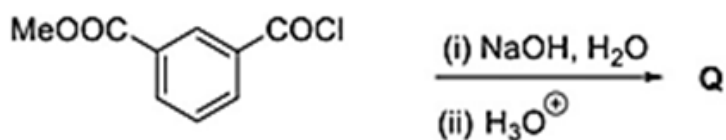
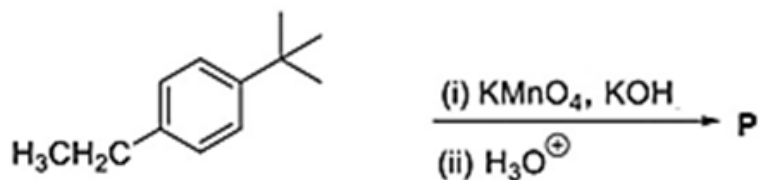
(A) 10                      (B) 8                      (C) 5                      (D) 7

3. Identify the correct chemical transformation into major organic product/s.



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4. Out of given four organic products P, Q, R and S which one is a compound of molecular formula  $C_8H_6O_4$ ?



- (A) Q, R and S      (B) P and R      (C) P only      (D) R and S

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## SECTION-2

This section consists of 3 Multiple Correct Answers Type Questions. Each question has 4 choices (A), (B), (C) and (D), out of which **ONE OR MORE THAN ONE CHOICE** is correct.

5. Select correct statements for electrochemical cells:

- (A) Such cells can be electrolytic or galvanic in nature
- (B) For spontaneous cell reactions  $\Delta G^\circ$  must be negative
- (C) Flow of current is always from cathode to anode via external circuit
- (D)  $E^\circ_{\text{AgBr}/\text{Ag}, \text{Br}^-_{\text{aq}}}$  is exactly equal to  $E^\circ_{\text{Ag}^+/\text{Ag(s)}}$

6. For a chemical reaction (I)  $2\text{R}_{(\text{g})} \rightleftharpoons \text{P}_{(\text{g})}$  the net rate of formation of  $\text{P}_{(\text{g})}$  is given as:

$$\frac{d[\text{P}]}{dt} = 2 \times 10^3 [\text{R}] - 5 \times 10^3 [\text{P}].$$

For another chemical reaction (II)  $2\text{X}_{(\text{g})} \rightleftharpoons 3\text{Y}_{(\text{g})}$  the net rate of formation of  $\text{Y}_{(\text{g})}$  is given as

$$\frac{d[\text{Y}]}{dt} = 6 \times 10^2 [\text{X}]^2 - 4.5 \times 10^3 [\text{Y}].$$

[Given that Arrhenius factors of all the gaseous reaction are equal]

Select correct statements

- (A) The ratio of forward rate constant i.e.  $k_{f(\text{I})} / k_{f(\text{II})}$  is  $10/3$ .
- (B) The ratio of backward rate constant i.e.  $k_{b(\text{I})} / k_{b(\text{II})}$  is  $10/3$ .
- (C) Both gaseous reaction i.e. I and II are exothermic.
- (D) Both gaseous reactions i.e. I and II are endothermic.

7. Select correct statements:

- (A) For a given non volatile, non-electrolyte solute boiling point and freezing point of the ideal solution shows a linear variation with molality
- (B) Addition of catalytic amount of  $\text{H}^+$  ions to an aqueous solution of sucrose shows a significant drop in freezing point of the solution
- (C) For an ideal solution of ethanol and n-propyl alcohol,  $\Delta H_{\text{mix}}$  is significantly negative.
- (D) Henry's law is NOT applicable for a gas at  $P_{\text{gas}} = 1 \text{ atm}$ , if at the given temperature Henry's constant  $K_{\text{H}}$  of the gas is 500 torr

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**SECTION-3**

**This section consists of 2 Paragraphs.** Based on each paragraph, there are **TWO (02)** questions. The answer to each question is a **NUMERICAL VALUE**. If the numerical value has more than two decimal places, **truncate/round-off** the value to **TWO** decimal places

**Paragraph for Questions 1 - 2**

Consider the given reaction mechanism for reaction between  $\text{H}_{2(g)}$  and  $\text{Br}_{2(g)}$  to form  $\text{HBr}_{(g)}$ .

**Step 1 :**  $\text{Br}_2 \rightleftharpoons 2\text{Br}^\cdot$   $E_f = 20.52 \text{ KJ / mole}$ ,  $E_b = 30.81 \text{ KJ / mole}$

**Step 2 :**  $\text{Br}^\cdot + \text{H}_2 \rightleftharpoons \text{HBr} + \text{H}^\cdot$   $E_f = 56.30 \text{ KJ / mole}$ ,  $E_b = 15 \text{ KJ / mole}$

**Step 3 :**  $\text{H}^\cdot + \text{Br}^\cdot \longrightarrow \text{HBr}$   $E_f = 0.15 \text{ KJ / mole}$

1. Based upon above mechanism the value of overall activation energy  $E_A$  for the reaction  $\text{H}_{2(g)} + \text{Br}_{2(g)} \longrightarrow 2\text{HBr}$  is \_\_\_\_\_. (in KJ/mole)
2. If concentration of both  $\text{H}_{2(g)}$  and  $\text{Br}_{2(g)}$  become four times of the original then ratio of new rate of the reaction and original rate of the reaction will be \_\_\_\_\_.

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**Paragraph for Questions 3 - 4**

$\text{Ti}_x\text{O}$  is a non stoichiometric compound formed due to metal deficiency defects in  $\text{TiO}$ . The metal can exist in two states,  $\text{Ti}^{2+}$  and  $\text{Ti}^{3+}$  to maintain electrical neutrality.

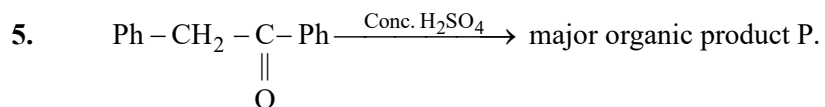
3. In  $\text{Ti}_{0.96}\text{O}$  percentage of the metal existing in +3 state is \_\_\_\_\_ .
4. If X-ray diffraction density of  $\text{Ti}_x\text{O}$  ( $x < 1$ ) is 5.35 g/cc while its pyknometer density is 5.20 g/cc then mass(g) of 1.2 mole sample of the  $\text{Ti}_x\text{O}$  is \_\_\_\_\_ .  
[Atomic masses : Ti = 45, O = 16]

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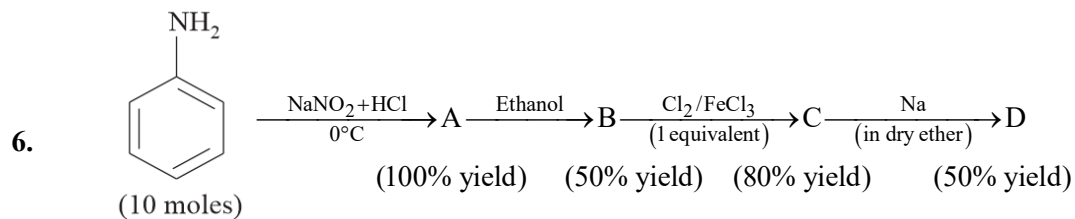
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**SECTION-4**

This section consists of 6 NON-NEGATIVE INTEGER Type Questions. The answer to each question is a NON-NEGATIVE INTEGER.



Maximum number of atoms in a plane for a molecule of product P would be \_\_\_\_\_.



Moles of organic product D produced in above reaction sequence is \_\_\_\_\_.

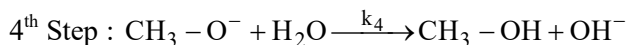
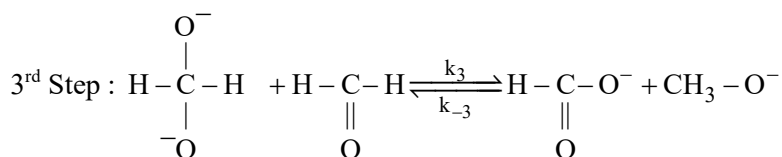
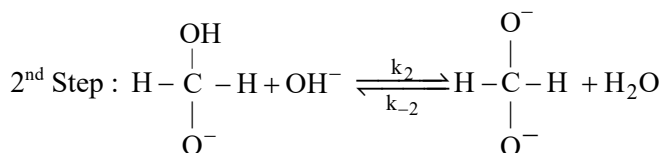
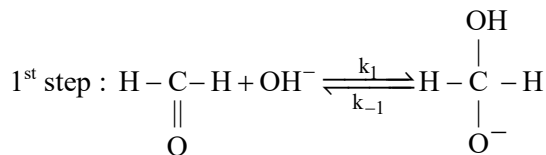
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7. As per the given mechanism the rate law for the reaction between  $\text{H}-\overset{\overset{\text{O}}{\parallel}}{\text{C}}-\text{H}$  and nearly 100% KOH is rate

$$r = k[\text{HCHO}]^x [\text{OH}^-]^y. \text{ Value of } x + y \text{ is } \underline{\hspace{2cm}}.$$

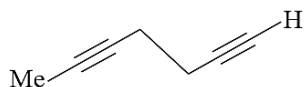
Mechanism:



8. 0.01M, 100 ml HCl was titrated with 0.01 M NaOH. If  $\lambda_{\text{H}^+}^\circ = 380 \text{ S cm}^2 \text{ mole}^{-1}$ ,  $\lambda_{\text{Na}^+}^\circ = 150 \text{ S cm}^2 \text{ mole}^{-1}$ ,  $\lambda_{\text{OH}^-}^\circ = 270 \text{ S cm}^2 \text{ mole}^{-1}$  and  $\lambda_{\text{Cl}^-}^\circ = 250 \text{ S cm}^2 \text{ mole}^{-1}$  and the  $\lambda^\circ$  values remain invariable during the reaction then value of resistance of the solution at the equivalent point is  $x\Omega$ . If cell constant is  $0.2 \text{ cm}^{-1}$ . Value of  $x$  is  $\underline{\hspace{2cm}}$ .

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9.



- (1) Na, Liquid  $\text{NH}_3$   
 (2)  $\text{CH}_3\text{CH}_2\text{Br}$   
 (3) Pd-BaSO<sub>4</sub>, Quinoline  
 (4) Cold dilute  $\text{KMnO}_4$ ,  $\text{OH}^-$

Number of stereoisomers of the organic product P that will be formed in above reaction is \_\_\_\_\_.

10. A first order reaction is found to take  $t_1$  minutes for 99.9% completion and  $t_2$  minutes for 50% completion. Find the value of  $\frac{t_1}{t_2}$ .

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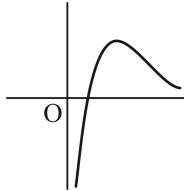
**SUBJECT III : MATHEMATICS**

**60 MARKS**

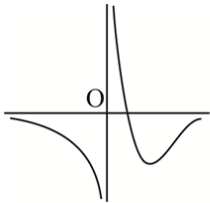
**SECTION-1**

This section consists of 4 Multiple Choice Questions. Each question has 4 choices (A), (B), (C) and (D), out of which **ONLY ONE CHOICE** is correct.

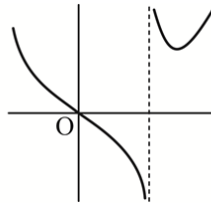
1. The graph of function  $f(x)$  is shown below :



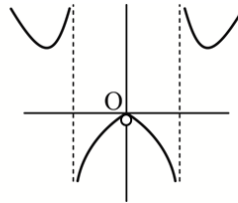
Then the graph of  $g(x) = \frac{1}{f(|x|)}$  is :



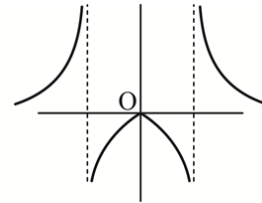
(A)



(B)



(C)



(D)

2. The value of  $\sec^2(\tan^{-1} 2) + \operatorname{cosec}^2(\cot^{-1} 3)$  is :

(A) 14

(B) 15

(C) 16

(D) 17

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3. Let  $g$  be the inverse function of a differentiable function  $f$  and  $G(x) = \frac{1}{g(x)}$ . If  $f(4) = 2$  and  $f'(4) = \frac{1}{16}$ , then the value of  $(G'(2))^2$  equals to :
- (A) 1                      (B) 4                      (C) 16                      (D) 64
4. Let  $f(x) = \begin{cases} 1 + \sin x, & x < 0 \\ x^2 - x + 1, & x \geq 0 \end{cases}$ , then :
- (A)  $f$  has a local maximum at  $x = 0$                       (B)  $f$  has a local minimum at  $x = 0$   
(C)  $f$  is increasing everywhere                      (D)  $f$  is decreasing everywhere
- 

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**SECTION-2**

This section consists of 3 Multiple Correct Answers Type Questions. Each question has 4 choices (A), (B), (C) and (D), out of which **ONE OR MORE THAN ONE CHOICE** is correct.

5. Which of the following is(are) incorrect ?
- (A) If  $f(x) = \sin x$  and  $g(x) = \ln x$  then range of  $g(f(x))$  is  $[-1, 1]$
- (B) If  $x^2 + ax + 9 > x \forall x \in R$  then  $-5 < a < 7$
- (C) If  $f(x) = (2011 - x^{2012})^{\frac{1}{2012}}$  then  $f(f(2)) = \frac{1}{2}$
- (D) The function  $f: R \rightarrow R$  defined as  $f(x) = \frac{x^2 + 4x + 30}{x^2 - 8x + 18}$  is not surjective
6. Let  $|f(x)| \leq \sin^2 x, \forall x \in R$ , then :
- (A)  $f(x)$  is continuous at  $x = 0$
- (B)  $f(x)$  is differentiable at  $x = 0$
- (C)  $f(x)$  is continuous but not differentiable at  $x = 0$
- (D)  $f(0) = 0$
7. If  $y = mx + 5$  is a tangent to the curve  $x^3 y^3 = ax^3 + by^3$  at  $P(1, 2)$ , then :
- (A)  $a + b = \frac{18}{5}$       (B)  $a > b$       (C)  $a < b$       (D)  $a + b = \frac{19}{5}$

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### SECTION-3

This section consists of 2 Paragraphs. Based on each paragraph, there are TWO (02) questions. The answer to each question is a **NUMERICAL VALUE**. If the numerical value has more than two decimal places, **truncate/round-off** the value to **TWO** decimal places

#### Paragraph for Questions 1 - 2

Let  $f: R \rightarrow R$  be differentiable function such that  $f(x) = x^2 + \int_0^x e^{-t} f(x-t) dt$ .

1. The value of  $\frac{12}{5} \int_0^1 f(x) dx$ .
2. If  $f(x) = \frac{x^{k_1}}{k_1} + x^{k_2}$ . Find  $k_1 + k_2$ .

#### Paragraph for Questions 3 - 4

Consider the two curves  $C_1: y = 1 + \cos x$  and  $C_2: y = 1 + \cos(x - \alpha)$  for  $\alpha \in \left(0, \frac{\pi}{2}\right)$ , where  $x \in [0, \pi]$ . Also the area of the figure bounded by the curves  $C_1, C_2$  and  $x = 0$  is same as that of the figure bounded by  $C_2, y = 1$  and  $x = \pi$ .

3. The value of  $[\alpha]$  is \_\_\_\_\_. ( $[\cdot]$  represents greatest integer function)
4. For the values of  $\alpha$ , area bounded by  $C_1, C_2, x = 0$  and  $x = \pi$  is \_\_\_\_\_.

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**SECTION-4**

This section consists of 6 NON-NEGATIVE INTEGER Type Questions. The answer to each question is a NON-NEGATIVE INTEGER.

5. If  $2 \tan^{-1} \frac{1}{5} - \sin^{-1} \frac{3}{5} = -\cos^{-1} \frac{63}{\lambda}$ , then  $\lambda =$
6. The value of  $x, x \in (2, \infty)$  where  $f(x) = \sqrt{x + \sqrt{8x - 16}} + \sqrt{x - \sqrt{8x - 16}}$  is not differentiable is \_\_\_\_\_.
7. Let set of all possible values of  $\lambda$  such that  $f(x) = e^{2x} - (\lambda + 1)e^x + 2x$  is monotonically increasing for  $\forall x \in R$  is  $(-\infty, k]$ . Find the value of  $k$ .
8. If the mean and the standard derivation of the data 3, 5, 7,  $\alpha$ ,  $\beta$  are 5 and 2 respectively, and  $\alpha$ ,  $\beta$  are roots of equation  $x^2 + ax + b = 0$ ; find  $a + b$ .
9. The value of  $\int \frac{\tan x}{\tan^2 x + \tan x + 1} dx = x - \frac{2}{\sqrt{A}} \tan^{-1} \left( \frac{2 \tan x + 1}{\sqrt{A}} \right) + C$   
Then the value of A is \_\_\_\_\_.
10. If  $\int \frac{dx}{\cos^3 x - \sin^3 x} = A \tan^{-1}(f(x)) + B \ln \left| \frac{\sqrt{2} + f(x)}{\sqrt{2} - f(x)} \right| + C$  where  $f(x) = \sin x + \cos x$  find the value of  $(12A + 9\sqrt{2}B) - 3$ .

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